



Molten Salt FP Release Tests in the GBI7 Experiment

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Grain Boundary Inventory 7 (GBI7) Experiment

- Last chance to use samples trace-reirradiated in NRU reactor
 - Trace-reirradiation technique was developed for GBI determination in UO_2
- Four different types of tests
 - GBI of MOX (similar to usual GBI of UO_2)
 - GBI of thoria (grinding; extension of previous work at Whiteshell Laboratories and elsewhere)
 - FP release from UO_2 under molten lead
 - FP release from molten salts
- Molten salt tests
 - 1000°C for about 4500 s – beyond design basis accident
 - Survey to determine whether FP release occurs
 - Inert, steam and air environments
 - preliminary results – data processing not complete



Molten Salt Preparation

- Trace-reirradiated UO_2 from NRU filler bundle
 - 38 kW peak linear power, 31 MWd/kgU burnup
 - Minimize segregation of noble metals into five-metal particles
- Grind to powder, mix with KHF_2 and NaF , add ~1% Zr
 - $\text{UO}_2 + 4 \text{KHF}_2 \rightarrow 2 \text{H}_2\text{O} + \text{K}_4\text{UF}_8$
 - Chvala studied reactor physics of 50.5%NaF – 21.5%KF – 28% UF_4
 - NaF lowers melting point
 - Zr consumes any excess F/HF from KHF_2 , and will generate some “ UF_3 ” in the mixture (desirable from reactor corrosion viewpoint)
- For some tests, substitute NaCl for NaF
 - Some fast reactor designs have chloride fuel salt separated from fluoride heat transfer salt
 - In order for significant FP release to occur, the fuel salt container must fail and the chloride and fluoride salts will mix

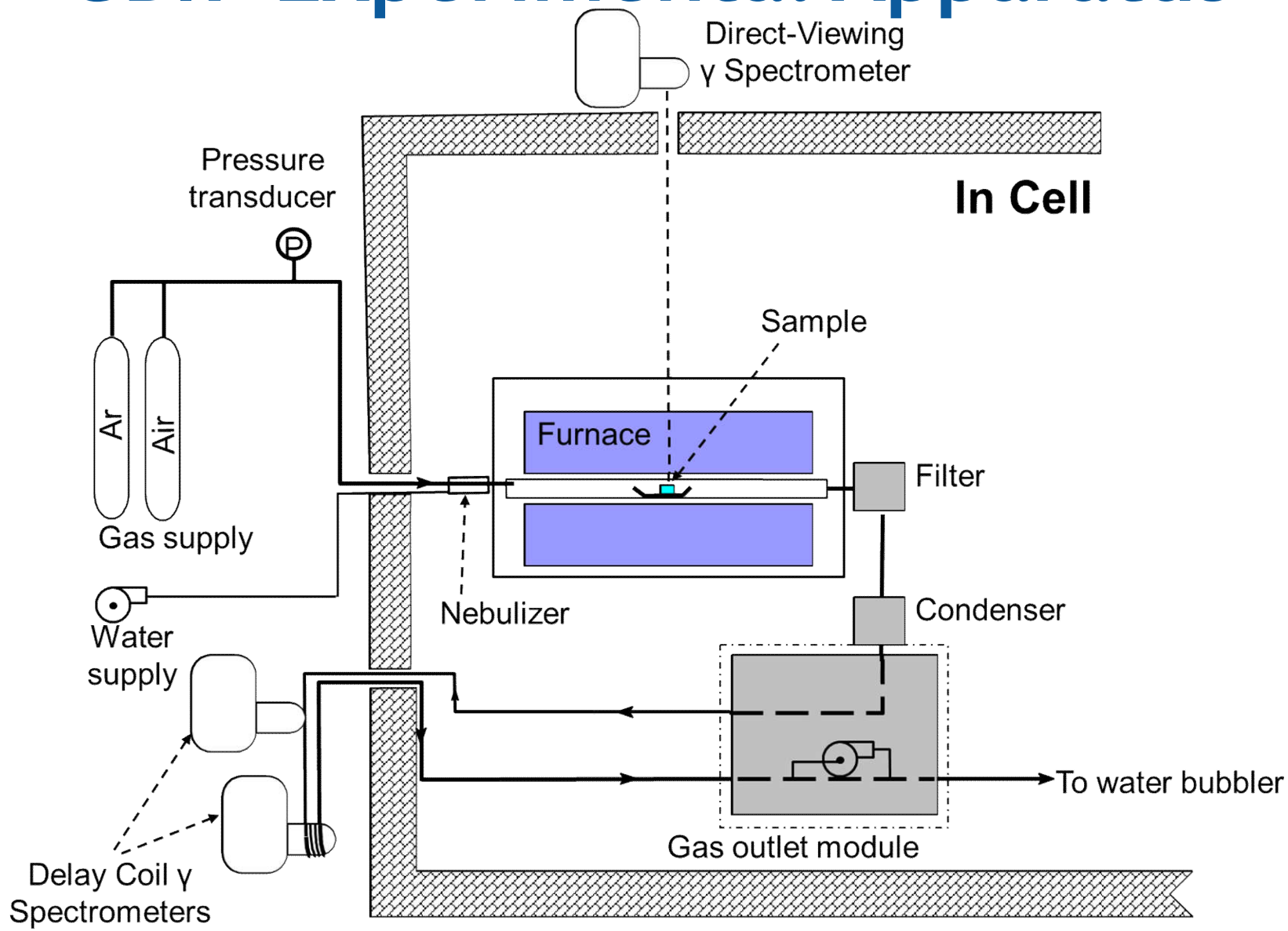


Molten Salt Test Technique

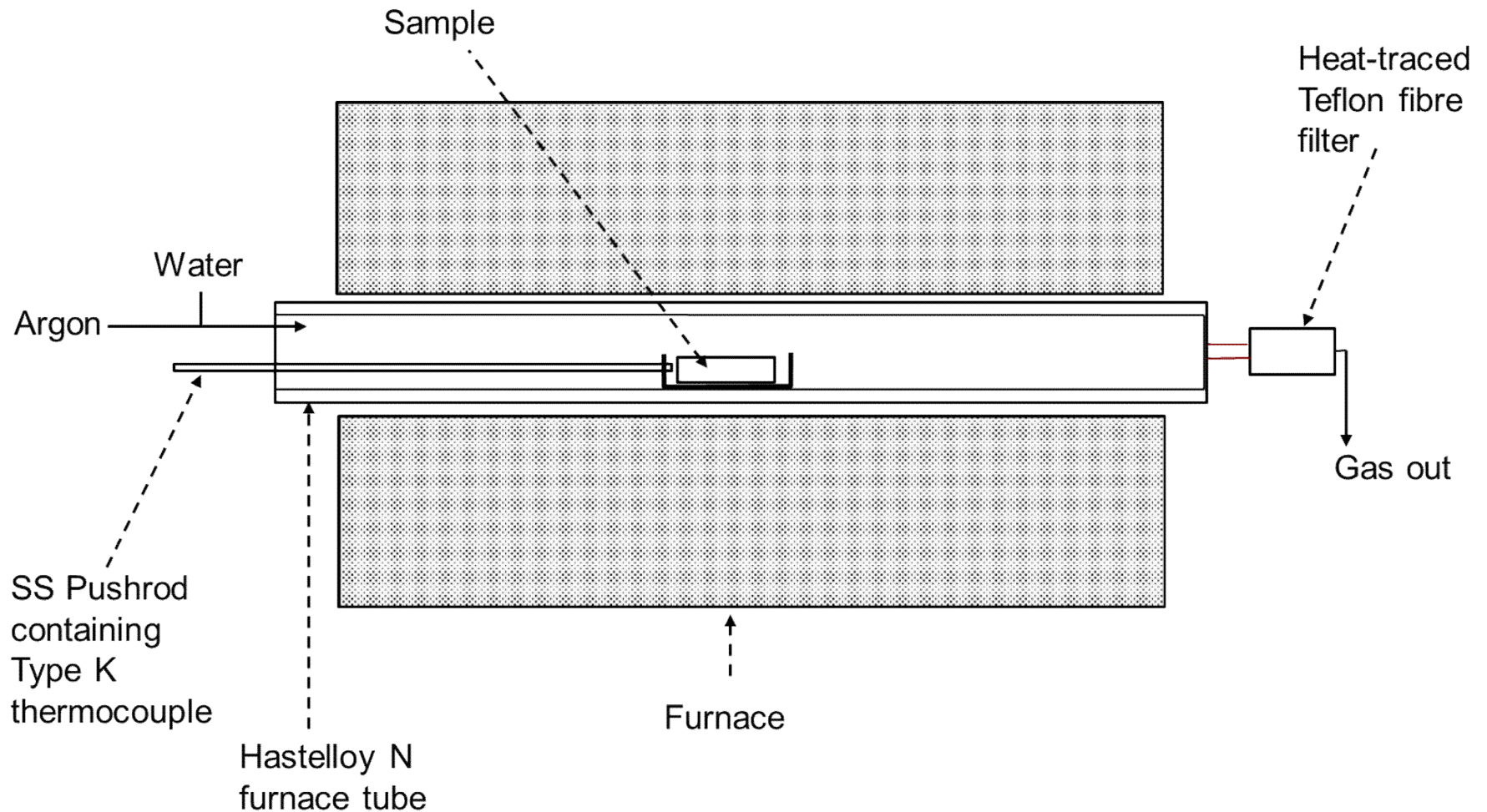
- Heat to $\sim 400\text{-}500^\circ\text{C}$ to cause reaction
 - Noble gases release during reaction – indicator for completeness of reaction
 - Minimal releases of other FP at this temperature
- Heat molten sample to 1000°C for about 4500 s
 - Observe using gamma spectrometry
 - Post-test analysis using SEM of cooled mixture



GBI7 Experimental Apparatus



GBI7 Experimental Apparatus



Tests

Test Name	Salt Type	Environment
MS1	UF ₄ -KF-NaF-1%Zr	Ar, 0.25 mmol/s
MS2	UF ₄ -KF-NaF-1%Zr	62% steam, 0.75 mmol/s
MS3	UF ₄ -KF-NaCl-1%Zr	69% steam, 0.67 mmol/s
MS4	UF ₄ -KF-NaCl-1%Zr	Ar, 0.25 mmol/s
MS5	UF ₄ -KF-NaF-1%Zr	air, 0.25 mmol/s
MS7	ThF ₄ -KF-NaF-1%Zr	Ar, 0.25 mmol/s



Preliminary results

- Fission product release at 1000°C
 - ~30% release of Cs isotopes
 - Comparable release of I-131
 - Much less I-131 release from chloride-fluoride salt than from fluoride-only salt
 - Release of Ru, particularly in oxidizing environment
- Fuel Behaviour
 - Inert-environment tests showed salt layer in bottom of boat
 - Oxidizing-environment tests showed salt “climbing over the walls” of the boat
 - Strong wetting of oxidized metal surface caused some salt to leave the boat



Future plans

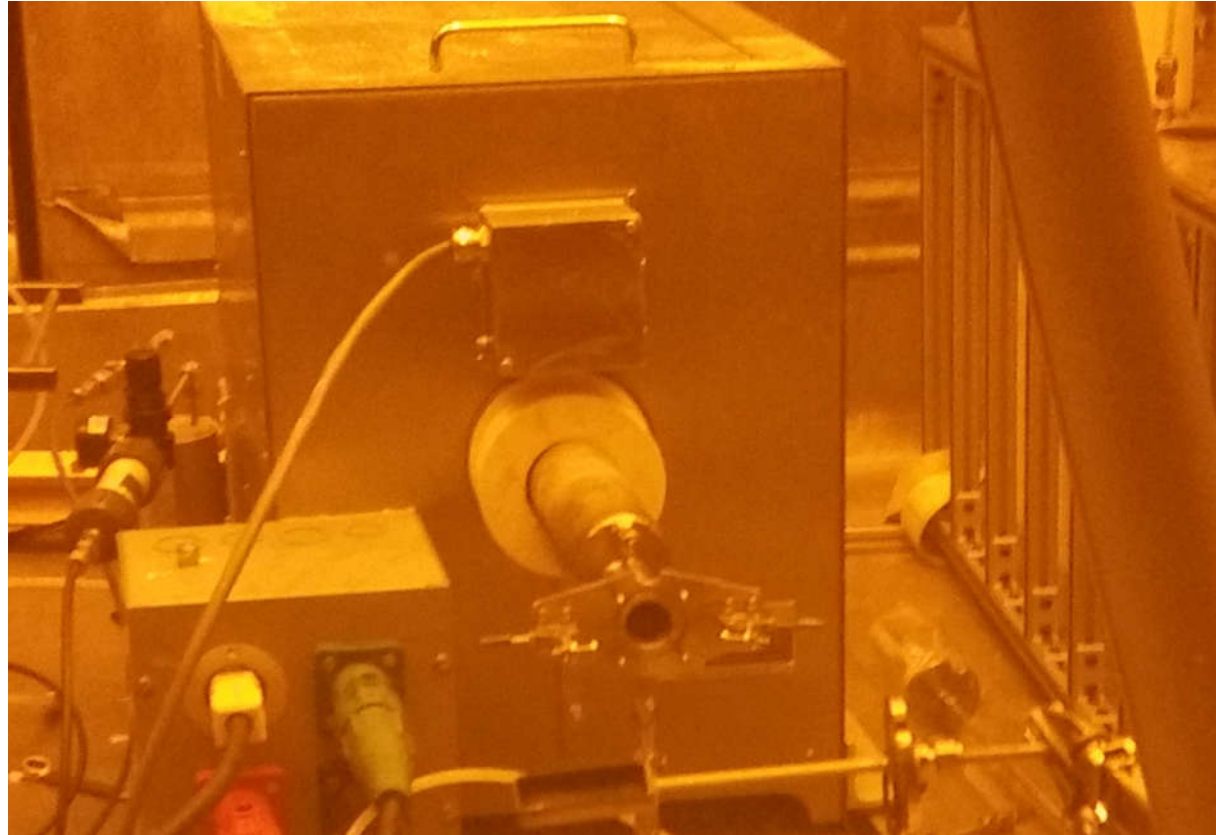
- Molten salt samples will be examined
 - Completeness of reaction with UO_2
 - Oxidation state
 - FP segregation
- Future tests
 - Wetting behaviour in oxidizing environment
 - FP release at lower temperatures (NOC and transient)
- Establish trace-reirradiation capabilities with another reactor
 - Shorter time between irradiation and test to examine behaviour of other FP (e.g., Te, Mo)



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Thank you. Merci.
Questions?

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